


*Methods, IPC, Weber*

<h2>EPR-Spectroscopy</h2> <p><i>Electron Paramagnetic Resonance</i></p>	<p>Model: <i>Bruker ESP380E</i>          Unit and Room: <i>Physical Chemistry, 5th floor, R. 503b</i>          Responsible: <i>Thomas Berthold (203-6212)</i>          Further information: <i><a href="http://www.physchem.uni-freiburg.de/akweber/forschung/eprfolder/index.html">http://www.physchem.uni-freiburg.de/akweber/forschung/eprfolder/index.html</a></i></p>
<p>Short Description:</p> <p>Continous-wave EPR-Spectrometer operating at X-band (9–10 GHz) and Q-band (34–36 GHz) microwave frequencies.</p>	<p>Picture of the Equipment</p> 
<p>Available Experiments/Techniques:</p> <p>continuous-wave EPR, including transient EPR (TREPR)</p>	
<p>Special Equipment:</p> <p>Low temperature unit (cryostat/resonator) for temperature range from 5 to 300 K.          Optical sample excitation (pulsed Nd:YAG/OPO laser system: 430–800 nm, 6 ns pulse length, &lt;10 Hz laser pulse repetition rate)          Goniometer for measurements of oriented samples (single crystals, liquid crystals)</p>	
<p>Measurements on the equipment are currently done by:</p>	<p><input type="checkbox"/> Students  <input type="checkbox"/> Students after Introduction  <input checked="" type="checkbox"/> Students after extensive training  <input checked="" type="checkbox"/> Trained scientific service personal</p>
<p>Recent Publications, where this instrument was important (optional): Give citation</p>	<p><i>Angew. Chem. Int. Ed.</i> 48 (2009) 404–407</p>
<p>Typical problems that may be solved with this instrument:</p>	<p>– <i>identification of radicals</i>          – <i>electronic structure determination of paramagnetic centers (organic radicals, transition metal ions, defect centers, optically excited states (triplets, radical pairs))</i>          – <i>measurement of short-lived (&gt;10 ns) paramagnetic intermediate states (triplet states, spin-polarized radical-pair states)</i></p>