


Methods, IAAC, Janiak

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| <p>Titration automats</p> <p><i>volumetric analysis</i></p> | <p>Model: Unit and Room: Responsible: Further information:</p> | <p><i>div. Metrohm titrinos + Schott titroline Chemie II, Zintl-lab Prof. C. Janiak, S. Zuelsdorf</i></p> |
| <p>Short Description:</p> <p>Automatic volumetric/titration analysis of liquid samples with potentiometric end point determination through ion-selective electrodes</p> | <p>Picture of the Equipment</p> | |
| <p>Available Experiments/Techniques:</p> <p>acid-base titration (e.g. total acid in juices and wine, carbonate for water hardness) precipitation titration (e.g. chloride with Ag+) complexometric titration (e.g. Ca and Mg in water, mineral water, milk with EDTA) all with potentiometric end point determination through ion-selective electrodes</p> |  | |
| <p>Special Equipment:</p> <ul style="list-style-type: none"> - normal pH-electrodes; - pH-solvotrode for acid-base titration in non-aqueous environment; - metal ion-selective electrodes for: Ag, Ca/Mg, Cu | | |
| <p>Measurements on the equipment are currently done by:</p> | <p><input type="checkbox"/> Students <input checked="" type="checkbox"/> Students after Introduction <input type="checkbox"/> Students after extensive training <input type="checkbox"/> Trained scientific service personal</p> | |
| <p>Recent Publications, where this instrument was important (optional): Give citation</p> | | |
| <p>Typical problems that may be solved with this instrument:</p> | <p><i>rapid, inexpensive and simple concentration determinations of various ionic species in the %- to upper ppm range</i></p> | |